

ALPHA-DECAY-INDUCED DAMAGE AND RECOVERY IN FUEL-LIKE ACTINIDE DIOXIDES

Yehuda Eyal

Department of Chemistry

Technion – Israel Institute of Technology, Haifa 32000, Israel

The significant structural stability of 12 $^{238}\text{PuO}_2$, $^{239}\text{PuO}_2$, $^{240}\text{PuO}_2$, $^{241}\text{AmO}_2$ and $^{244}\text{CmO}_2$ samples and 14 fuel-like (U,Pu)O₂, (U, ^{241}Am)O₂, (Pu, ^{241}Am)O₂ and (Pu, ^{244}Cm)O₂ samples under self α decay is explored by a radiation annealing model [1]. At ambient temperature, following exposure of ~ 0.2 dpa (displacements per atom), lattice expansion is limited to $\Delta V/V = 0.0090 \pm 0.0009$. This behavior indicates limited accumulation of Ac (actinide) and O Fps (Frenkel pairs), and thereby suggests the importance of a steady-state of damage and recovery.

In our model, event-by-event production of Ac and O Fps is combined with prompt removal of randomly-occurring closely-spaced interstitial-vacancy pairs. Recombination is controlled by critical annihilation volumes, v_{ac} and v_o . Guided defect mobility is generated by electronic and mechanical distortions. The fractional volume expansion is given by $\Delta V/V = [\Omega_{ac}/v_{ac} + \Omega_o/v_o] \exp(-w\lambda Nt)$. Here, Ω_{ac} and Ω_o represent volume increases per surviving Ac and O Fp, N is the density of the Ac atoms, λ is the weighted average decay constant of an atom, and t is the exposure time. This expression reproduces all experimental data.

Our MD simulations on UO₂ at $< 5\text{K}$ [2], and two more recent similar computations [3,4], reveal that interstitial (or vacancy) in the O sublattice recombine promptly with its 1st or 2nd closest vacancy (or interstitial). Therefore, the joint crystal volume that is occupied by a stable pair of non-overlapping O interstitial and vacancy is $v_o = 1.218 \text{ nm}^3$. Thus, apparently the density of surviving O Fps at defect saturation is $v_o^{-1} = 0.82 \text{ nm}^{-3}$. Some close U interstitials and vacancies also recombine promptly. However, there is evidence that U Fps are efficiently eliminated via thermal diffusion. Thus, our model yields $\Omega_{\theta} = 0.0110 \pm 0.0018 \text{ nm}^3$. This result coincides with high values of this parameter as derived by static computations [5], and is close to 0.0137 nm^3 as derived by the elastic continuum model [6].

Briefly, the mean volume damaged per α -decay event is $w = 490 \pm 80 \text{ nm}^3$ ($\sim 3,000$ unit cells). The simple inverse exponential dependence of $\Delta V/V$ on time indicates that defect saturation is achieved even in a first-exposed lattice region. Under any large number of Fps that are created per α -decay event, assumed $\sim 1,500$, about 400 O Fps survive.

We find that the stability of the AcO₂ lattice is independent of the composition, chemical properties and the decay half-lives of the actinide cations. These properties are an advantage in the context of safe use of advanced complex AcO₂ reactor fuels as well as direct disposal of burnt reactor fuels. Long-term stability is assured by the presence of ancient highly crystalline uraninite and thorianite [7].

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